Buddha Institute Of Technology					STITUT OF THE		
Department: Mechanical Engineering							
Academic Semester: July- Dec 2022					BIT	<u> </u>	
Semester: V	Section: B	Course Code: KME501		Course	Transfer		
Course Instructor:	Contact Hours /week: 5 (4+1) # of credits		# of credits: 04				
CIE Marks: 50 SEE Marks:10			)		Exam Hours	: 03	

Prerequisites if	any:						
Code No	Course Name	Description	Semester				
NOT APPLICABLE							

Content delivery: Chalk & Board, DLP, System/Laptop with social media videos

COURSE SYLLABUS:						
Module No	Contents of Module	Hrs	COs			
1	Thermodynamics and Heat Transfer. Modes of Heat Transfer: Conduction, convection and radiation. Effect of temperature on thermal conductivity of materials, Introduction to combined heat transfer mechanism, General differential heat conduction equation in the rectangular, cylindrical and spherical coordinate systems. Initial and boundary conditions, Simple and Composite Systems in rectangular, cylindrical and spherical coordinates with and without energy generation, Concept of thermal resistance, Analogy between heat and electricity flow, Thermal contact resistance and over all heat transfer coefficient; Critical radius of insulation.	18	C01			
2	Heat transfer from extended surfaces, Fins of uniform cross-sectional area, Errors of measurement of temperature in thermometer wells, Transient heat conduction, Lumped capacitance method, Time constant, Unsteady state heat conduction in one dimension only, Heisler charts.	16	CO2			
3	Basic concepts, Hydrodynamic boundary layer, Thermal boundary layer, Approximate integral boundary layer analysis, Analogy between momentum and heat transfer in turbulent flow over a flat surface, Mixed boundary layer, Flow over a flat plate, Flow across a single cylinder and a sphere, Flow inside ducts, Thermal entrance region, Empirical heat transfer relations, Relation between fluid friction and heat transfer, Liquid metal heat transfer. Physical mechanism of natural convection; Buoyant force, Empirical heat transfer relations for natural convection over vertical planes and cylinders, horizontal plates and cylinders, and sphere, Combined free and forced convection.	14	CO3			
4	Basic radiation concepts, Radiation properties of surfaces, Black body radiation Planck's law, Wein's displacement law, Stefan Boltzmann law, Kirchoff's law, Gray body, Shape factor, Black-body radiation, Radiation exchange between diffuse non black bodies in an enclosure, Radiation shields, Radiation combined with conduction and convection; Absorption and emission in gaseous medium; Solar	9	CO4			

	radiation, Green house effect.		
5	Types of heat exchangers, Fouling factors; Overall heat transfer coefficient, Logarithmic mean temperature difference (LMTD) method, Effectiveness-NTU method, Compact heat exchangers, Introduction to condensation phenomena, Heat transfer relations for laminar film condensation on vertical surfaces and on outside & inside of a horizontal tube, Effect of non-condensable gases, Drop wise Condensation, Heat pipes, Boiling modes, pool boiling, Hysteresis in boiling curve, Forced convection boiling.	9	CO5

#### COURSE OUTCOMES: At the end of the Course, the Student will be able to:

KME501.1	understand the mode of heat transfer and governing equations concerned to conduction and one dimensional steady state heat flow
KME501.2	express the knowledge of design skills of heat exchangers
KME501.3	describe the real time applications of fluid medium heat transfer
KME501.4	illustrate the real time applications of radiation mode of heat transfer
KME501.5	analyse the heat exchanger and process of boiling, condensation and mass transfer

### Mapping of CO v/s PO:

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
KME501.1	3	-	-	-	-		-	-	-	-	-	-
KME501.2	3	-	-	-	-		-	-	-	-	-	-
KME501.3	2	-	-	-	-		-	-	-	-	-	-
KME501.4		-	2	-	-	2	-	-	-	-	-	-
KME501.5		-	-	3	-	1	-	-	-	-	-	-

**Correlation levels:** 1-Slight (Low)

2-Moderate (Medium)

3-Substantial (High)

Mapping of CO v/s PSO:

	PSO1	PSO2
KME501.1	1	-
KME501.2	1	-
KME501.3	1	-
KME501.4	1	-
KME501.5	1	-

Gap in the syllabus	NIL

Topics	to	be	covered	Demographic survey – A case study
beyond	sylla	ibus		

Assessment Methodologies:

Sl. No.	Description	Туре
1	Student Assignment	Direct
2	Internal assessment	Direct
3	University exam	Direct
4	Student feedback	Indirect
5	5 Alumni feedback	
6	Employers feedback	Indirect

### LESSON PLAN

Lecture #	Module#	Topics	RBT	Course	Planned	Actual	Faculty	Remarks
			Levels	Mapping	Date	Date	Sign	
1.		Thermodynamics and Heat			22/08/2022			
		Transfer,			22/00/2022			
2.		Modes of Heat Transfer			23/08/2022			
		Conduction			23/00/2022			
3.		Convection, Radiation,			24/08/2022			
4.		Effect of temperature on						
		thermal conductivity of			26/08/2022			
		materials.						
5.		Tutorial			27/08/2022			
6.		Introduction to combined			29/08/2022			
		heat transfer mechanism.						
7.	1	General differential heat	13	CO1	30/08/2022			
	1	conduction equation in the	LJ	COI				
		rectangular coordinate						
		systems						
8.		General differential heat			31/08/2022			
		conduction equation in the						
		rectangular coordinate						
		systems						
9.		General differential heat			02/09/2022			
		conduction equation in the						
		cylindrical coordinate						
		systems						
10.		Tutorial			03/09/2022			

11		General differential heat			05/09/2022		
11.		conduction equation in the			05/07/2022		
		conduction equation in the					
		cymurical coordinate					
		systems			0.6/0.0/2022		
12.		General differential heat			06/09/2022		
		conduction equation in the					
		spherical coordinate					
		systems.					
13.		General differential heat			07/09/2022		
		conduction equation in the					
		spherical coordinate					
		systems. Initial and					
		boundary conditions					
14		Simple and Composite			09/09/2022		
14.		Systems in rectangular			07/07/2022		
		aulindrical and spherical					
		cynnuncar and sphericar					
		coordinates with and without					
		energy generation.					
15.		Tutorial			10/09/2022		
16.		Concept of thermal			12/09/2022		
		resistance, Analogy between					
		heat and electricity flow.					
17.		Thermal contact resistance			13/09/2022		
		and over all heat transfer					
		coefficient;					
18.		Critical radius of insulation.			14/09/2022		
19.		Basic radiation concepts.			16/09/2022		
		Radiation properties of					
		surfaces					
20		Tutorial			17/09/2022		
20.		Black body radiation			23/00/2022		
21.		Planck's law			23/09/2022		
22		Wein's displacement law					
22.		Stefan Boltzmann law			24/09/2022		
22		Kirchoff's law Gray body			26/00/2022		
23.		Shape feater Black hady			20/09/2022		
24.		radiation			21/09/2022		
25	4	Tutorial	L4	CO4	28/00/2022		
<u> </u>					28/09/2022		
20.		Radiation exchange between			30/09/2022		
		diffuse non black bodies in					
		an enclosure.			0.1.11.0.120.000		
27.		Radiation shields, Radiation			01/10/2022		
		combined with conduction					
		and convection.					
28.		Absorption and emission in			03/10/2022		
		gaseous medium, Solar					
		Radiation, Green house					
		effect.					
29.		Types of heat exchangers,			07/10/2022		
30.		Tutorial			08/10/2022		
31.	1	Fouling factors. Overall heat			10/10/2022		
	5	transfer coefficient.	L4	CO5			
32	-	Logarithmic mean			11/10/2022		
		temperature difference			11,10,2022		
		(I MTD) method					
		(Lint D) memou.					

33.		Effectiveness-NTU method.			12/10/2022	
34.		Compact heat exchangers.			14/10/2022	
		Introduction to condensation				
		phenomena,				
35.		Tutorial			15/10/2022	
36.		Heat transfer relations for			17/10/2022	
		laminar film condensation				
		on vertical surfaces and on				
		outside & inside of a	L4	CO5		
		horizontal tube.				
37.		Effect of non-condensable			18/10/2022	
		gases, Dropwise				
		Condensation.Heat pipes,				
38.		Boiling modes, pool boiling.			19/10/2022	
39.		Hysteresis in boiling curve,			21/10/2022	
		Forced convection boiling.				
40.	-	Tutorial			22/10/2022	
41.		Basic concepts,			31/10/2022	
		Hydrodynamic boundary				
42		The second secon			01/11/2022	
42.		A normal boundary layer,			01/11/2022	
		Approximate integral				
12		A palagy between			02/11/2022	
43.		momentum and heat transfer			02/11/2022	
		in turbulent flow over a flat	L4	CO3		
		surface				
44.		Mixed boundary layer, Flow			04/11/2022	
		over a flat plate.			•	
45.		Tutorial			07/11/2022	
46.		Flow across a single			08/11/2022	
		cylinder, Flow across a				
		single sphere.				
47.		Empirical heat transfer			09/11/2022	
		relations, Relation between				
	3	fluid friction and heat				
		transfer.				
48.		Liquid metal heat transfer,			14/11/2022	
		Physical mechanism of				
		natural convection;			15/11/2022	
49.		Buoyant force. Empirical			15/11/2022	
		neat transfer relations for				
		vertical planes and cylinders				
50		Tutorial			16/11/2022	
50.		Flow inside ducts Thermal			18/11/2022	
51.		entrance region			10/11/2022	
52	1	Empirical heat transfer			19/11/2022	
		relations for natural				
		convection over horizontal				
		plates and cylinders.				
53.	1	Empirical heat transfer			21/11/2022	
		relations for natural				
		convection over sphere.				
54.		Combined free and forced			22/11/2022	

		convection.					
55.		Tutorial			23/11/2022		
56.		Heat transfer from extended			25/11/2022		
		surfaces.					
57.		Fins of uniform cross-			26/11/2022		
		sectional area.					
58.		Fins of uniform cross-			28/11/2022		
		sectional area.					
59.		Errors of measurement of			29/11/2022		
		temperature in thermometer					
		wells.					
60.		Tutorial			31/11/2022		
61.		Transient heat conduction,			04/12/2022		
62.		Lumped capacitance			05/12/2022		
	2	method.	L6	CO2			
63.		Lumped capacitance			06/12/2022		
		method.					
64.		Time constant			07/12/2022		
65.		Tutorial			12/12/2022		
66.		Time constant,			13/12/2022		
67.		Unsteady state heat			13/12/2022		
		conduction in one dimension					
		only.					
68.		Unsteady state heat			13/12/2022		
		conduction in one dimension					
		only.					
69.		Heisler charts.			13/12/2022		
70.		Heisler charts.			13/12/2022		

# \*L1 – Remembering; L2 – Understanding; L3 – Applying; L4 – Analysing; L5 – Evaluating; L6 - Creating

## Literature:

Reference 1	R1	Heat And Mass Transfer By Er. R.K. Rajput	11 <sup>th</sup>	S.Chand Company	2017(81-219-2617-3)
Reference 2	R2	A Textbook of Heat and Mass Transfer By R.C. Rchdeva	5 <sup>th</sup>	New age international limited publishers	2011(978-81-224- 2785-1)

### Sample Questions:

Question	Questions
1	What do you understand by overall heat transfer coefficient?
2.	Explain black body, white body, gray body and opaque body.
3.	How heat exchangers are classified?
4.	For a steady flow process from state 1 to state 2 enthalpy changes from 400 KJ/kg to 100 KJ/kg
	and entropy changes from 1.1 KJ/kg-k to 0.7 KJ/kg-k and the ambient temperature is 300 K. Find
	the change in availability.(GATE 2009)
5.	What do you understand by Inversion curve? Define Joule Thomson Co-efficient with diagram.
6.	A reversible engine is supplied 900 KJ of heat from a heat source at 500 K. The engine develops
	300 KJ of net work and rejects heat to two heat sinks at 400 K and 300 K. Determine the engine
	thermal efficiency and magnitude of heat interaction with each of the sink.
7.	Determine the enthalpy, Specific Volume, internal energy and entropy of superheated steam at 15
	bar pressure and 220 $^{0}$ C. Take specific heat of superheat equal to 2.2 KJ/Kg-K
8.	Three Carnot engines 1, 2 and 3 operate between temperature of 800 K and 500 K. Make
	calculations for the intermediate temperatures if the work produced by the engines are in the ratio
	of 5:3:1
9.	Discuss the coefficient of volume expansion, adiabatic and isothermal
	compressibility. Also find the loss in available energy due to given heat transfer. If 3 kg of
	gas ( $C_V = 0.81$ kJ/kg K) initially at 2.5 bar and 400 K receives 600 kJ of near from an
	infinite source at 1200 K and the surrounding temperature is 290 K
10.	Derive Steady Flow Energy Equation (S.F.E.E.). Also write the steady flow energy equation for
	turbine, boiler and centrifugal pump
11.	Discuss the Clapeyron equation and also explain the Joule-Kelvin effect with help of
	inversion curve and inversion temperature.
12.	Determine the enthalpy, Specific Volume, internal energy and entropy of superheated steam at 25
	bar pressure and 250 °C. Take specific heat of superheat equal to 2.12 KJ/Kg-K.
12	Provide the second of the last of the fraction of the second
13.	Expain the concept of principle of entropy increase
14.	Steam at 1000 kPa and 500°C enters an engine and expands to 20 kPa. If the exhaust steam has a drypers fraction of $0.0^{\circ}$ make calculation for the drop in enthalpy and change in entropy
15	A Single stage air turbine is to operate with an inlet pressure and temperature of 8 bar and 600K
15.	The outlet pressure and temperature are 1 bar and 300K the surroundings pressure and
	temperature is 1 bar and 200K. Mass flow rate is 5 kg/s, determine the decrease in availability, the
	maximum work and the Irreversibility. For air take $C_P = 1.005$ KJ/Kg-k and R = 0.287KJ/Kg-K
16.	A gas initially at 1.5 bar pressure, 0.15 m <sup>3</sup> volume and 300 K was compressed polytropically
	(pv <sup>1.48</sup> = constant) to 13 bar pressure. Determine the change in entropy. Also work out the
	approximate entropy change obtained by dividing the heat transferred by the mean absolute
	temperature during compression. Take $C_p=1.04$ KJ/Kg-K and $C_v=0.745$ KJ/Kg-K
17.	Define the various methods of refrigeration and application of the refrigeration.
18.	Explain the Rankine cycle with the help of neat sketch, P-V and T-S diagram. If 5 kg water at 45°C
	is heated at a constant pressure of 10 bar until it becomes superheated vapour at 300°C. Find the
	change in volume, enthalpy, internal energy and entropy
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19.	Explain the vapour compression refrigeration cycle and its C.O.P. with the help of T-S, P-H and
	flow diagram. Can this cycle be reversible?
20.	An air- water vapour mixture at 25 $^{0}$ C. and 1 bar has a relative humidity of 75 percent. Determine a) The partial pressure of water vapour and dry air. b) the specific volume of each constituent c) the dew point temperature, the specific humidity and the saturation ratio. d) the density of the mixture.

Assessment rubrics that is going to be adopted for direct attainment is depicted in below table

Level of Achievement	Elaboration on Course Grading Description	Bench Mark Set (Out of 100)
Excellent (A)	The Student's performance is outstanding in almost all the intended course learning outcomes	21 to 25
Good (B)	The student's performance is good in most of the intended course learning outcomes.	15 to 20
Marginal (C)	The student's performance is barely satisfactory. It marginally meets the intended course learning outcomes	12 to 14
Fail (F)	The Students performance is inadequate. Student fails to meet many of the intended course learning outcomes	Less than 12